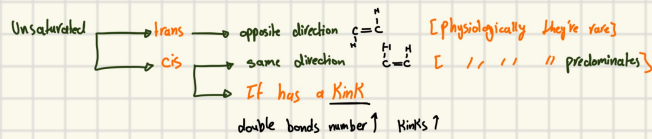


Fatty Acids:

- Aliphatic monocarboxylic acid
- Formula $R-(CH_2)_n-COOH$
- Vary in unsaturated level [saturated fats are rich in electrons cuz they're reduced]
- Vary in length
 - physiological range [12-24]
 - Abundant [16-18]
- Amphipathic molecule [hydrophilic + hydrophobic]

- Functions
 - Building blocks for other lipids
 - Fuel molecules
 - derivatives of cellular molecules
 - Modification of proteins [lipoprotein]

- Classification depending on saturation
 - Saturated → all C-C bonds are single
 - Unsaturated
 - monounsaturated → it has **one** double bond between its carbons
 - polyunsaturated → it has **two or more** bond between its carbons



- Melting point depends on
 - chain length [increasing carbon atoms number increase the interactions, more surface area to interact with]
 - saturation degree [Saturated > monounsaturated > polyunsaturated]
- Tight packing → **high** melting point
Kinks → loose packing → **low** melting point.

- Saturated fatty acids
 - Short F.A [2-4]
 - liquid in nature
 - volatile at RT
 - water soluble [the hydrophilic portion is **dominant**]
 - Acetic acid [vinegar], butyric acid [butter]
 - Medium F.A [6-10]
 - Solid at RT
 - non-volatile at RT
 - water **soluble**
 - capric, caprylic, caproic acids [Goat's milk]
 - Long F.A [12-20]
 - Solid at RT
 - non-volatile
 - water **insoluble**
 - Palmitic acid [palm oil], stearic acid

- Unsaturated F.A
 - palmitoleic acid hexadecenoic acid, w7, 16:1, Δ^9
 - oleic acid Octadecenoic acid, w9, 18:1, Δ^9
 - Linoleic acid Octadecadienoic acid, w6, 18:2, $\Delta^9, 12$
 - Linolenic acid Octadecatrienoic acid, w3, 18:3, $\Delta^{9, 12, 15}$
 - Arachidonic acid Eicosatetraenoic acid, w6, 20:4, $\Delta^{5, 8, 11, 14}$
 - Eicosapentaenoic acid, w3, 20:5, $\Delta^{5, 8, 11, 14, 17}$

- Saturated acid: Myristic acid : Tetradecanoic acid

- The last methyl group is called ω [omega], we consider it the 1st carbon in naming in the **Omega style**

- Examples
- $\omega 3$
 - linolenic acid
 - EPA, Eicosapentaenoic acid 20:5
 - DHA, Docosahexaenoic acid 22:6, $\Delta^{4, 7, 10, 13, 16, 19}$
 - $\omega 6$
 - Linoleic acid
 - Arachidonic acid

* Linoleic acid and Linolenic acid are Essential FA [From diet only]

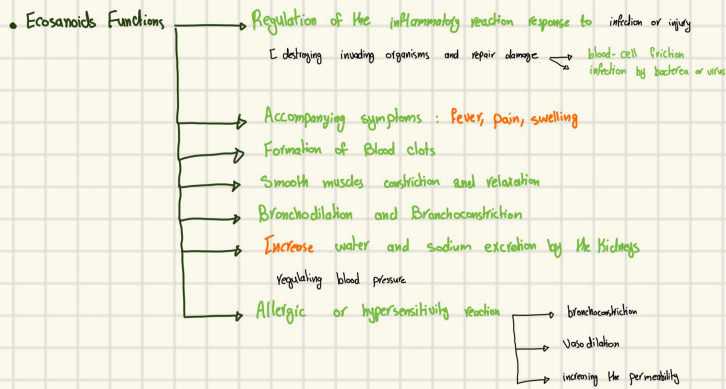
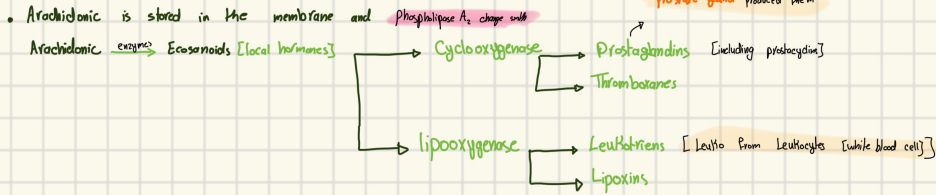
Linoleic acid is precursor of Arachidonates

Linolenic acid is precursor of DHA, EPA

Derived fatty acid

- Wax : - Solid simple lipids
 - It is formed by
 - monohydric alcohol [molecular weight higher than glycerol [16-36], ex. palmitoyl alcohol
 - long chain fatty acid [14-36]
 - Water insoluble [hydrophobic]
 - Not easily hydrolyzed, indigestible
 - Very resistant to rancidity
 - No nutritional value
 - External coating layer on plants leaves to prevent water loss

- Linoleic acid
 1. Elongation
 2. Desaturation
 → Arachidonic acid



Specific Functions



- Aspirin causes GI ulcers [eating COX1 produces a layer that protects the stomach and GI tract from the acids, etc. and Aspirin prevents the COX1 production. As a consequence the acids with their enzymes will irritate the tissues leading to ulcers]

Extra Notes :

- Penicillin is the first antibiotic discovered
- Statin related to cholesterol and cancer therapy
- Glucophage + Metformin for diabetic and prediabetic patients
 - + It is good for people over 45 years
- GLP1 related to weight loss ex. Wegovy, Ozempic, etc

- In humans there are 3 forms of the COX, COX1, COX2, COX3
- Aspirin prevents both, even though it should prevent only COX2.

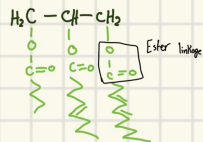
COX1 → Homeostatic function such as : GI and Renal Vasc., platelet aggregation, Macrophages differentiation

COX2 → Inflammation

- Aspirin is good but it can cause **excessive bleeding** especially for elderly
 - Celecoxib
- Celebrex** → blocks COX2 only, but it can cause heart failure, etc.
- Vioxx** was from the same family but it was **withdrawn** from the market cuz it caused heart failure
- Omega 3**
 - **linolenic acid** → reduces inflammatory reaction
 - **EPA, DHA**
- Omega 6** → **linoleic acid, Arachidonic**
- Omega 9** → **Oleic acid** → reduces cholesterol in circulation

Triacylglycerol

- Storage lipid consist of glycerol and 3 fatty acids.
- It forms **adipose tissue** and it is found also in **liver**

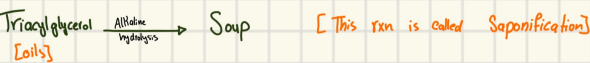


- Glycerol + 1 fatty acid = **Monoglyceride**
- Glycerol + 2 fatty acid = **Di glyceride**
- Glycerol + 3 fatty acid = **Triglyceride**

- Simple Triacylglycerol**: identical fatty acids
- Mixed Triglyceride**: different fatty acids

- Vegetable oils consist almost mainly of **unsaturated fatty acids** → kinks → low melting point → **liquid**
- Animal fats have **high percentage** of **saturated fatty acids** → tightly packed → high melting point → **solid**

- Saponification**:
 - hydrolysis under alkaline condition [**Alkaline hydrolysis**]
 - hydrolysis agents
 - **steam, acid** [chemical agents]
 - **Enzyme** example [lipase of pancreas], [biological agents]
 - produces salts of fatty acids, release the glycerol.
 - The product [**Soaps**] cause **emulsification** of oily materials.



- we react the oil with strong base (NaOH) to produce RCOONa fatty acid salt.

How Soap works?

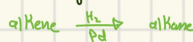
- Hydrophobic tails cluster together (oriented inside) to create a **nonpolar** microenvironment and the **hydrophilic** heads [The COO⁻ part of the fatty acid] interacts with water.
- The resulting spherical clusters are called **Micelles**
- Dirt and grease will be trapped in the hydrophobic interaction [**Emicelles**] and the whole complex can easily be rinsed away

[The water pulls away the hydrophobic head with its micelles together]



Trans Fat

- Hydrogenation**: C-C double bonds in vegetable oils [unsaturated fats] can be hydrogenated in order to yield saturated fats



- Even though animal fats are unhealthy, they have some benefits
 - better cooking properties
 - better taste

- Companies use method to convert unsaturated vegetable oil to **solid** form by **partial hydrogenating** them

• Partial hydrogenation converts some of the double bonds, **NOT All**, to single bonds generating **Trans Fats**

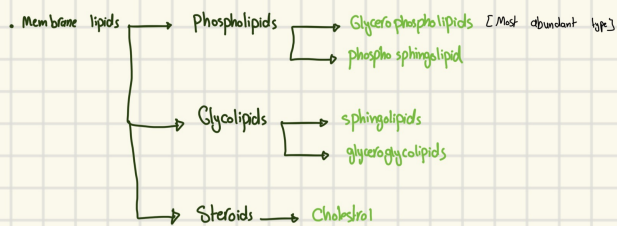
• Trans Fat is loved by industrial companies for
 → Its appealing taste
 → It has longer shelf life than [cis, or saturated fatty acids]

Why is it bad for health → Our body can't handle it cuz it's a foreign molecule and it accumulates in the blood vessels which will cause an elevation **risk** of the **coronary heart diseases**.
 [It also increases the LDL [bad cholesterol level] and decreases the HDL [Good Cholesterol] level which will lead generally to increase the cholesterol in the blood vessels, Furthermore, it will accumulate and leads to many abnormal conditions in the vessels] **Obese**

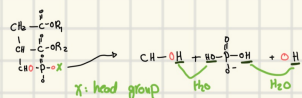
• Example on **Trans Fat** : - Margarine

- only about 9% of the C-C double bonds in starting material is hydrogenated.
- It will be soft in the fridge, and can melt on a warm toast easily.

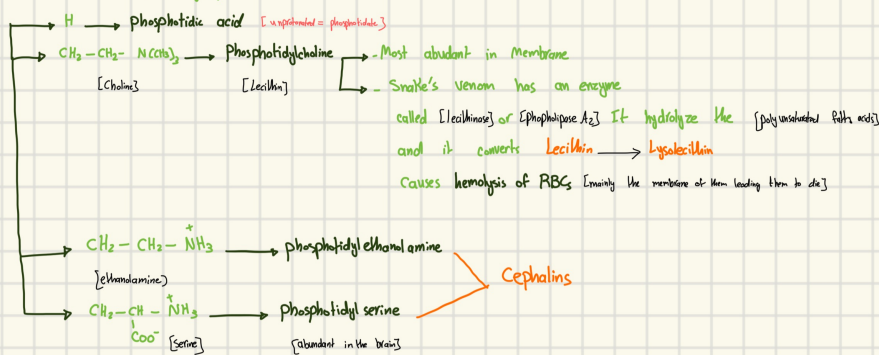
Membrane lipids



• Phosphoglycerol



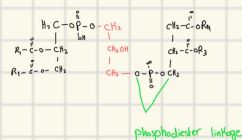
• Head Groups



• Lecithin is found in [lecithin, choline, inositol, etc.] because It can act as an **emulsifying agents**; meaning their amphipathic nature can allow them to surround **non polar** molecules and keeps them in suspension in water. [disolving fat in water]

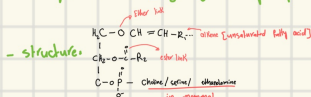
• Cardiolipins : - It's type of phosphoglycerol

- Found in the inner membrane of mitochondria [crista]
- Initially It was isolated from heart muscle [Cardiolipin]
- Diphosphatidyl-glycerol
- 4 fatty acids, 3 glycerol, 2 phosphate groups



• Plasmalogens :

- Type of glycerophospholipids
- Found in cell membrane fraction of **brain, liver, muscle, semen**
- They have protective role against **reactive oxygen species**
- Its precursor is Dehydroacetone phosphate

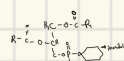


- Types



• Phosphatidyl inositol : - Cyclic sugar alcohol [invariant]

- Source: from the brain
- Structure: Glycerol, unsaturated FA, saturated FA, Phosphoric acid and inositol



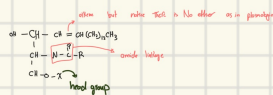
- Function
 - major component in cell membrane
 - signalling molecules produced upon hydrolysis [It is broken into active signaling molecules]

• Structure of Phospholipids

- Liposome [used to deliver drugs to the cells, endocytosis]
- micelle [as in soap, emulsification]
- bilayer sheet [as in cell membrane]



- Sphingolipids:
 - Found in the plasma membrane of all eukaryotic
 - Is highest in the central nervous system
 - The core is the long chain of amino alcohol, sphingosine

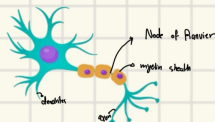


• Types of sphingolipids

- Ceramide [the head group would be H]
 - It's the simplest sphingolipid
- Sphingomyelin [the head group is phosphocholine]
 - It's the major component of the coating around nerve fiber
 - It's hydrophobic to prevent the electrons from escaping
- Glycosphingolipid [Glycolipids]
 - The only phosphosphingolipid

• Myelin

- composed of sphingomyelin
- Multiple sclerosis; immune cells attack the myelin and damage it.



Glycolipids

- Glycolipids have backbone either
 - Glycerol → Phosphatidyl-inositol
 - sphingosine
 - Cerebroside
 - Globoside
 - Ganglioside

- Sphingosine can attach carbohydrates on C1; And they're called Glycolipid
- Glycolipids are in the cell membrane acting as cell surface receptor [pathogen] and chemical messenger.

- Glycolipids
 - Cerebroside → simplest glycolipid; one hexose sugar [glucose, galactose] is attached to the sphingosin by O-glycosidic linkage.
 - Globoside → two or more sugars [glucose, galactose, N-acetylglucosamine]
 - Ganglioside → three or more sugars including at least one sialic acid [neuraminic acid]. They can bind to cholera toxins in the intestines and it facilitates its endocytosis to the cell [causing excessive diarrhea]

- Sulfatides:
 - They are synthesized from galactocerebroside with attachment of $[SO_3^-]$ on C3 [it gives it a -ve charge]
 - abundant in the brain myelin

- Sphingolipids can serve as intercellular communication and as antigenic determinant of the ABO blood groups.

- Some are used as receptors by viruses or bacterial toxins.
- The sugar sequence of the ABO antigen can bind to
 - protein [glycoprotein]
 - sphingosine [glycolipid, specifically [globoside]]

- Lipo protein
 - large structure, [emulsifies] having glycerophospholipids and proteins
 - Functions: transport different types of lipids [phospholipid, cholesterol, cholesteryl ester, triglycerides] in the blood plasma
 - Types
 - Chylomicrons
 - VLDL
 - LDL
 - IDL
 - HDL

* They differ in Density, size, proteins attached to it [collecting its role]

Density depends on **Proteins**
 Density ↑ size ↓ **Lipids**

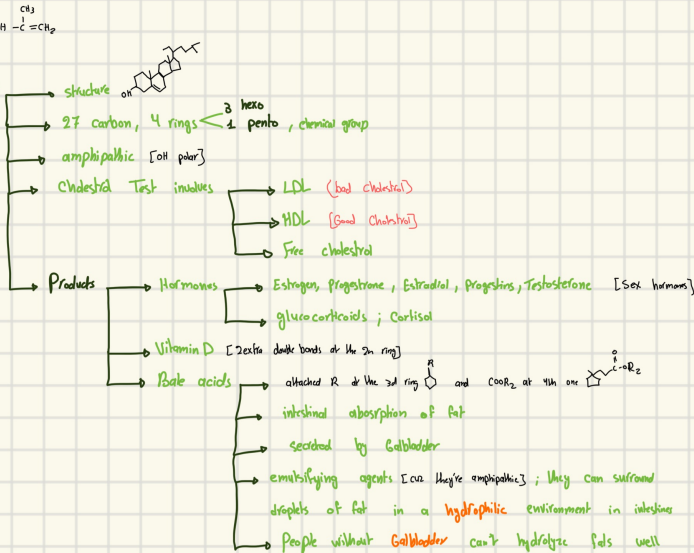
- When consumed fats reaches intestine; it'll get hydrolyzed and packaged into **Chylomicrons**. [absorption pathway, exogenous, dietary lipid]
- As **Chylomicrons** flow in the blood endothelial enzymes will dig the **fatty acids** to tissues to be metabolized, and chylomicrons are left carrying **cholesterol** to the liver.

Liver packages [cholesterol, fatty acids, lipids, etc] in **VLDL**; and as it flows in the blood the fatty acids from the triacylglycerol are removed; leaving cholesterol and increasing the density (↑ $\frac{\text{protein}}{\text{fat}}$), so it'll be converted into **LDL**; which will deliver **cholesterol** from the liver to peripheral tissues.

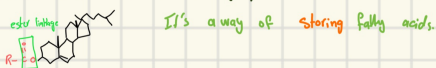
Liver also releases **HDL**; which transports **cholesterol** from the peripheral tissues to the liver

- Liver main function
 - Storing molecules [glucose, cholesterol, triacylglycerol, etc]
 - Detoxification

Steroids: - It's precursor **isoprene** $\text{CH}_2=\text{CH}-\text{C}(\text{CH}_3)_2$
 - It's nucleus  18 carbons
 - Most common example: **cholesterol**



cholesterol can bind to a fatty acid at [OH on C3] becoming a **cholesterol ester**; it becomes more hydrophobic and it loses its amphipathic feature.

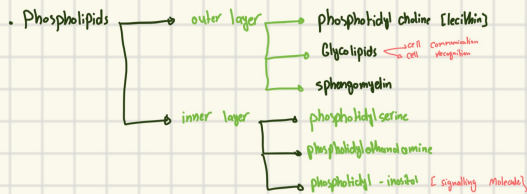


Bad cholesterol **LDL** can accumulate in the blood vessels causing → narrow blood vessel → closure

Atherosclerosis

Cell Membrane

- The plasma membrane is called **fluid mosaic model**
- 45% lipids, 45% proteins, 10% carbohydrates
- They exist side by side without forming other intermediate



cholesterol is found in both layers.

It is related to mammals X [plants, plants etc.]

It's distributed **EQUALLY**

But **Phospholipid Distribution** isn't uniform since each layer has different types.

- Saturated fatty acids have straight tails allow them to closely packed to each other [rigid membrane]
- Unsaturated fatty acids have kinks due to the double bonds; preventing tight packaging + increase fluidity.

cholesterol + cis unsaturated fatty acids prevents tight packing of hydrophobic chains → 1d the membrane liquid and pushes to move laterally in the plane of the leaflet → dynamic membrane [for signaling]

cholesterol can also stabilize the fluidic membrane by increasing hydrophobic interactions between fatty acids and cholesterol ring structure.

Unsaturated levels differ between nations:

- Africa: saturated ↑
- EsKimo: unsaturated ↑

 Their diet depends on fish [omega 3]